

IN THE SPECIFICATION

Please replace the title of the disclosure with the following new title:

-- LITHIUM SECONDARY BATTERY CONTAINING ORGANIC PEROXIDE IN
NON-AQUEOUS ELECTROLYTE, POSITIVE ELECTRODE, OR NEGATIVE ELECTRODE

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Please amend paragraph [0030] of the disclosure as follows:

Among them, phases comprising silicon and at least one transition metal selected from the group consisting of Ti, Ni, Co, Fe and Cu, alloy phases comprising silicon and the above-mentioned at least one transition metal, and intermetallic compound phases comprising silicon and the above-mentioned at least one transition metal are preferred. Exemplary phases comprising silicon and at least one transition metal selected from Ti, Ni, Co, Fe, Cu, Zr, and the like include: solid solution phases comprising silicon and at least one transition metal element selected from the group consisting of Ti, Ni, Co, Fe and Cu; and alloy phases comprising silicon and at least one intermetallic compound selected from the group consisting of $TiSi_2$, $TiSi$, $CoSi_2$, $CoSi$, $FeSi_2$, $FeSi$, $NiSi_2$, $NiSi$, and Cu_3Si .

In this case, it is further preferred that the above-mentioned at least one transition metal include Ti. Also, among these phases, phases comprising $TiSi_2$ are particularly preferred. This is because $TiSi_2$ has a high electronic conductivity.

It should be noted that a mixture of the above-mentioned material, such as alloy, with carbon material, such as graphite, may be used as the negative electrode active material unless the feature of high capacity is [[not]] impaired.

Please insert the following Table 1 at paragraph [0076]. This Table 1 is fully disclosed in PCT/JP2005/004443, of which the present application is a national phase.

[Table 1]

	Positive electrode active material	Negative electrode active material	Organic peroxide	Battery capacity (mAh)	Capacity retention rate (%)
Battery A1	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	Cumene peroxide	2635	79
Battery A2	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	m-chloroperbenzoic acid	2622	85
Battery A3	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	Diethyl peroxide	2611	76
Battery A4	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	α -cumyl peroxyneodecanate	2629	74
Battery A5	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	Benzoyl Peroxide	2623	77
Battery A6	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	1,1-di-t-butylperoxy cyclohexane	2590	67
Battery A7	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	Acetyl acetone peroxide	2636	72
Battery A8	LiCo _{0.98} Mg _{0.02} O ₂	Si thin film	Cumene peroxide	2610	78
Battery A9	LiCo _{0.98} Mg _{0.02} O ₂	Si thin film	m-chloroperbenzoic acid	2725	82
Battery A10	LiCoO ₂	Ti-Si alloy	m-chloroperbenzoic acid	2635	83
Battery A11	Li _{1-x} SCo _x O ₂	Ti-Si alloy	m-chloroperbenzoic acid	2810	80
Battery A12	LiNi _{0.8} Co _{0.15} Mn _{0.05} O ₂	Ti-Si alloy	m-chloroperbenzoic acid	2802	83
Battery A13	LiNi _{1/3} Co _{1/3} Mn _{1/3} O ₂	Ti-Si alloy	m-chloroperbenzoic acid	2585	85
Compara. battery 1	LiCo _{0.98} Mg _{0.02} O ₂	Ti-Si alloy	None	2635	57
Compara. battery 2	LiCo _{0.98} Mg _{0.02} O ₂	Si thin film	None	2750	55
Compara. battery 3	LiCo _{0.98} Mg _{0.02} O ₂	Graphite	m-chloroperbenzoic acid	2250	82
Compara. battery 4	LiCo _{0.98} Mg _{0.02} O ₂	Graphite	None	2284	80
Compara. battery 5	LiMn ₂ O ₄	Ti-Si alloy	m-chloroperbenzoic acid	2250	73
Compara. battery 6	LiMn ₂ O ₄	Graphite	m-chloroperbenzoic acid	1923	79